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## Structure Reports

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## 2,4-Diphenyl-2-(trimethylsilyloxy)-3-dihydro-2H,5H-pyrano[3,2-c][1]benzopyran-5-one

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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.047$
$w R$ factor $=0.139$
Data-to-parameter ratio $=14.2$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]In the title compound, $\mathrm{C}_{27} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si}$, the coumarin ring system is essentially planar and the pyran ring adopts a half-chair conformation.

## Comment

The recognition of the synthetic potential of heterocyclic quinone methides has aroused considerable interest in their generation and subsequent transformations. (Nair et al., 2001). We have recently investigated the cycloaddition reaction between coumarin quinone methide, 3-phenylmethylene- 2 H -1-benzopyran-2,4(3H)-dione and trimethyl[(1-phenylethenyl)oxy]silane, which gave the title compound, (I), as one of the products. An X-ray crystallographic analysis was undertaken to establish its structure and conformation.

(I)

In the title structure (Fig. 1), the bond lengths and angles are within normal ranges (Allen et al., 1987) and are in good agreement with the related compound 2-methyl-2-(trimethyl silyloxy)-3,4-dihydro-2H,5H-pyrano[3,2-c][1]benzopyran-5one (Peng et al., 2005). No classical intermolecular hydrogen bonds were found in the crystal structure but there is an intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen-bonding interaction (Table 2). The dihedral angle between the heterocyclic plane (C19-C21/O4/C22/C27) and the fused benzene ring is $4.74^{\circ}$. The pyran ring (C7-C9/C19/C20/O2) adopts a half-chair conformation, with atom C7 deviating from the C9/C19/C20/ O2 plane by $0.596 \AA$.

## Experimental

The title compound, (I), was prepared by the reaction of 4hydroxycoumarin, 3 equivalents of trimethyl[(1-phenylethenyl)oxy]silane and an excess of benzaldehyde. The mixture in dry dioxane was refluxed for about 18 h . Compound (I) was isolated by column chromatography of the reaction mixture on silica gel after evapora-

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tion of the solvent, in $31 \%$ yield. Single crystals of (I) were obtained by slow evaporation of a petroleum ether-ethyl acetate ( $3: 1 \mathrm{v} / \mathrm{v}$ ) solution.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{27} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Si} \\
& M_{r}=442.57 \\
& \text { Monoclinic, } P 2_{4} / n \\
& a=12.223(2) \AA \AA \\
& b=12.740(3) \AA \\
& c=15.113(3) \AA \\
& \beta=91.59(3)^{\circ} \AA \\
& V==2352.5(8) \AA^{3} \\
& Z=4
\end{aligned}
$$

$$
D_{x}=1.250 \mathrm{Mg} \mathrm{~m}^{-3}
$$

Mo K $\alpha$ radiation
Cell parameters from 25
reflections
$\theta=10-13^{\circ}$
$\mu=0.13 \mathrm{~mm}^{-1}$
$T=298$ (2) K
Block, colorless
$0.30 \times 0.20 \times 0.20 \mathrm{~mm}$

## Data collection

Enraf-Nonius CAD-4
diffractometer
$\omega / 2 \theta$ scans
Absorption correction: $\psi$ scan
(XCAD4; Harms \& Wocadlo, 1995)
$T_{\text {min }}=0.962, T_{\text {max }}=0.973$
4333 measured reflections
4129 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.047$
$w R\left(F^{2}\right)=0.139$
$S=1.00$
4129 reflections
290 parameters
H -atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.08 P)^{2}\right. \\
& +0.0023 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\text {max }}=0.007 \text { 。 } \\
& \Delta \rho_{\text {max }}=0.26 \text { e } \AA^{-3} \\
& \Delta \rho_{\min }=-0.24 \mathrm{e} \AA^{-3} \\
& \text { Extinction correction: SHELXL97 } \\
& \text { Extinction coefficient: } 0.0219 \text { (18) }
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right.$ ).

| $\mathrm{Si}-\mathrm{O} 1$ | $1.6648(16)$ | $\mathrm{Si}-\mathrm{C} 10$ | $1.846(3)$ |
| :--- | ---: | :--- | ---: |
|  |  |  |  |
| $\mathrm{O} 1-\mathrm{Si}-\mathrm{C} 10$ | $101.86(11)$ | $\mathrm{C} 19-\mathrm{O} 2-\mathrm{C} 9$ | $119.15(16)$ |
| $\mathrm{C} 9-\mathrm{O} 1-\mathrm{Si}$ | $131.77(14)$ | $\mathrm{C} 9-\mathrm{C} 8-\mathrm{C} 7$ | $113.06(17)$ |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 7-\mathrm{C} 20$ | $-2.4(3)$ | $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9-\mathrm{O} 2$ | $-55.1(2)$ |

Table 2
Hydrogen-bond geometry ( $\AA{ }^{\circ},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 14-\mathrm{H} 14 A \cdots \mathrm{O} 2$ | 0.93 | 2.38 | $2.708(3)$ | 100 |



Figure 1
The structure of (I), showing $30 \%$ probability displacement ellipsoids and the atom-numbering scheme.

The H atoms were positioned geometrically and were refined as riding, with $\mathrm{C}-\mathrm{H}=0.93-0.98 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.2 U_{\mathrm{eq}}(\mathrm{C})$ or $1.5 U_{\text {eq }}$ (methyl C).

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: XCAD4 (Harms \& Wocadlo, 1995); program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

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